

Metal Oxide Nano-Array Catalysts for Low Temperature Diesel Oxidation

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Project ID #: ACE095

Project Overview

Overall objective:

—To develop a unique class of cost-effective and high performance metal oxide based nano-array catalysts for low temperature (at 150 °C or lower) diesel oxidation

Timeline

- Project start date: 10/01/2014
- Project end date: 12/31/2016
- Percent complete: < 20%

Budget

- Total project funding
 - DOE share: \$1,450,000
 - Contractor share: \$380,139

Barriers

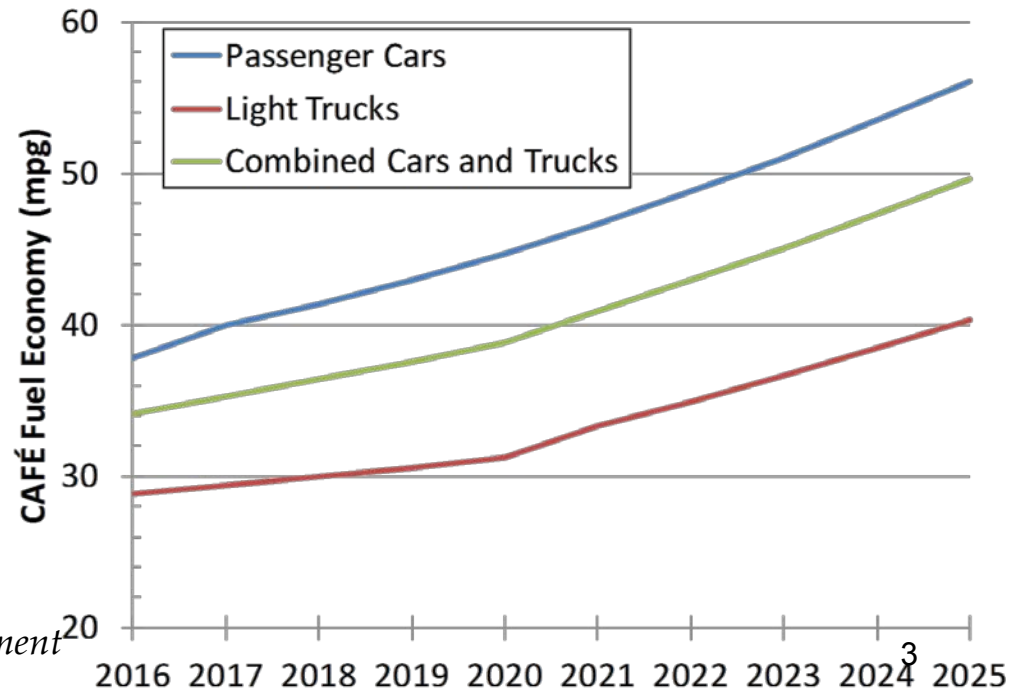
- Barriers addressed
 - Initial Formulation of nano-array catalysts
 - Assembly of nano-array catalysts with reduced usage of metal oxide and noble metals
 - CO and HCs oxidation tests at low temperature

Team Partners

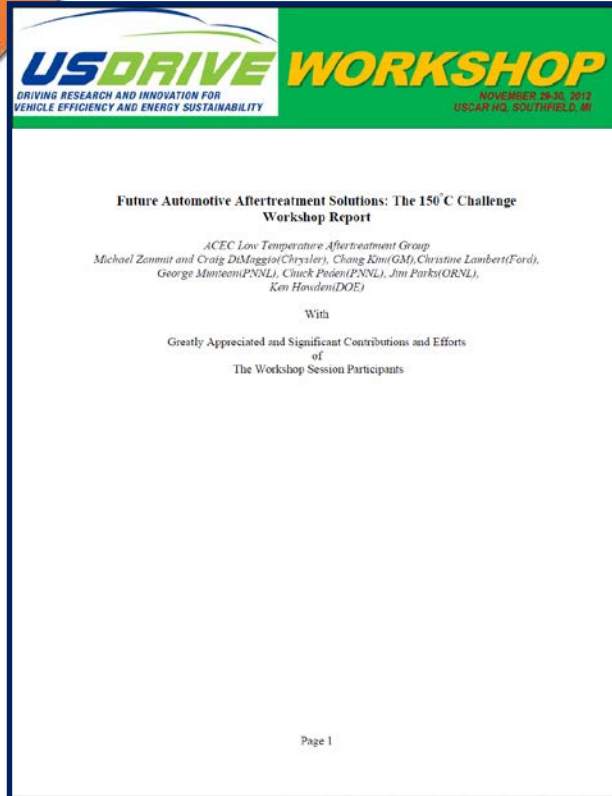
- ORNL, Umicore, and 3D Array Technology LLC

Project Relevance

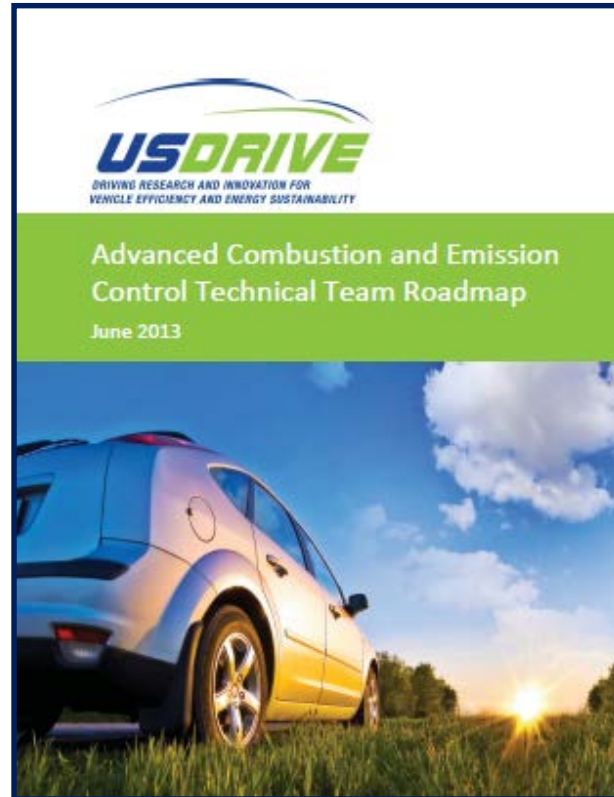
- Identify and formulate cost-effective and high performance nano-array based catalysts that are compatible to **low temperature combustion engines** with greater fuel efficiency and consequently lower exhaust temperature conditions
- Low temperatures catalysis challenges:
 - Emissions standards harder to meet
 - 10x higher HCs and CO, new chemistry at low temperature, need new DOC.
- Investigate nano-array based catalysts to improve low Temperature catalysis for emission control
 - ~90% conversion at 150°C or lower
- Fuel economy demands



Project Relevance



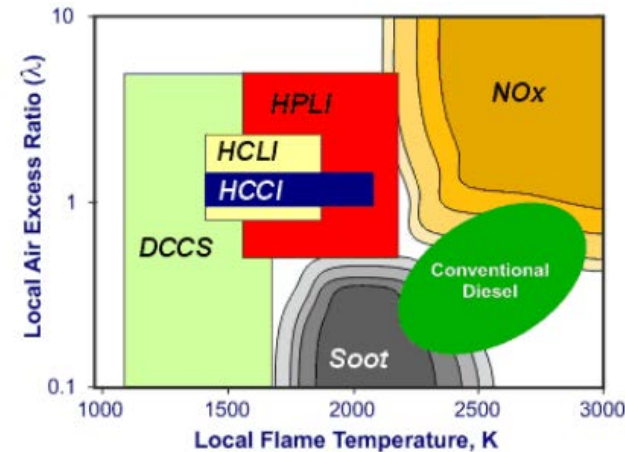
**USDRIVE “The 150°C Challenge”
Workshop Report**



**2013 USDRIVE ACEC Tech
Team Roadmap**

Needs addressed in this project:

- Lower temperature CO oxidation; HC oxidation; and NO_x reduction
- Reduced PGM
- Better thermal stability



W. Addy Majewski, Hannu Jääskeläinen, Engine Design for low emission, Dieselnets

Advanced combustion engine technologies:

- Low Temperature Combustion (LTC)
- Dilute Gasoline Combustion
- Clean Diesel Combustion (CDC)

Tasks and Approaches

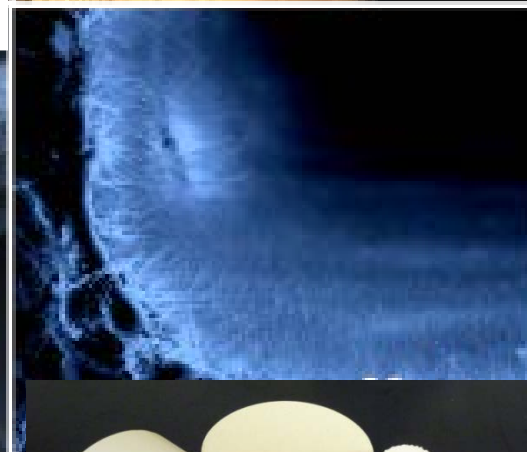
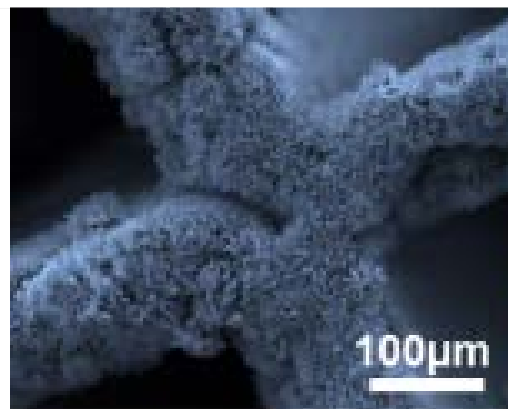
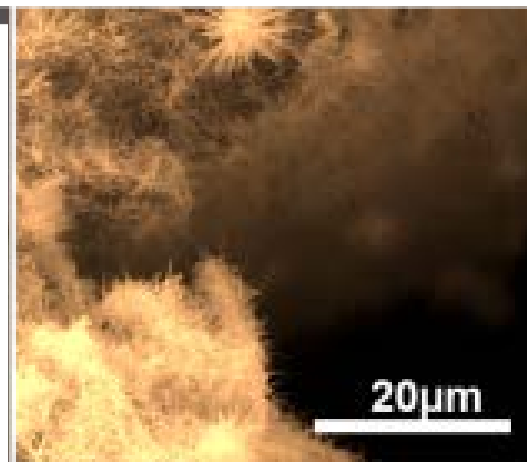
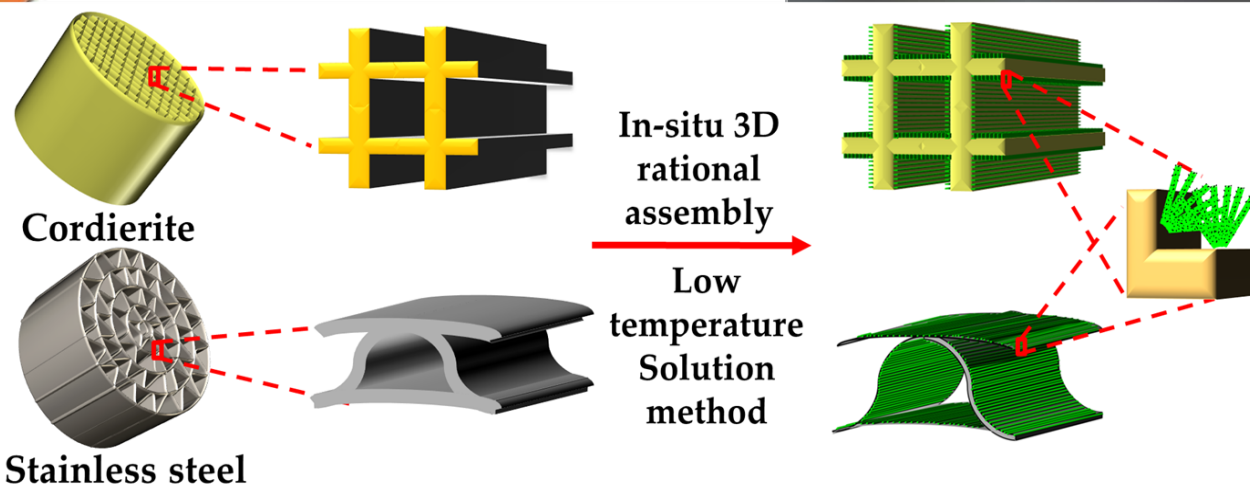
- Tasks in Quarters 1-2, 10/1/2014-3/31/2015
 - Design, assembly and characterization of metal oxide nano-array catalysts
 - Initiation of perovskite and noble metal loading
 - Testing of CO and HCs oxidation over nano-array catalysts
- Approaches:
 - ❑ **Synthesis and assembly:**

Directly grow and assemble 3D metal oxide nano-array catalysts on honeycomb substrates by in-situ solution and gas phase approaches.
 - ❑ **Nano-characterization:**

Investigate the structure, morphology, chemical properties of nano-array catalysts using a range of microscopy and spectroscopy techniques.
 - ❑ **Low Temperature Activity, and Stability:**

Explore the catalytic behavior and stability using benchtop reactors, thermal analysis and temperature programmed analysis tools.

In-situ Growth of Nano-arrays onto Honeycomb Monoliths



- In-situ growth of nano-array on monolith
- Free of binders, robustness due to the strong substrate-array adhesion after in-situ growth
- Reduced PGM and other materials usage
- Improved efficiency due to size, shape, and structure

Ren, Gao et al., *Angew. Chem. Int. Ed.*, **2014**, 53(28), 7223–7227.

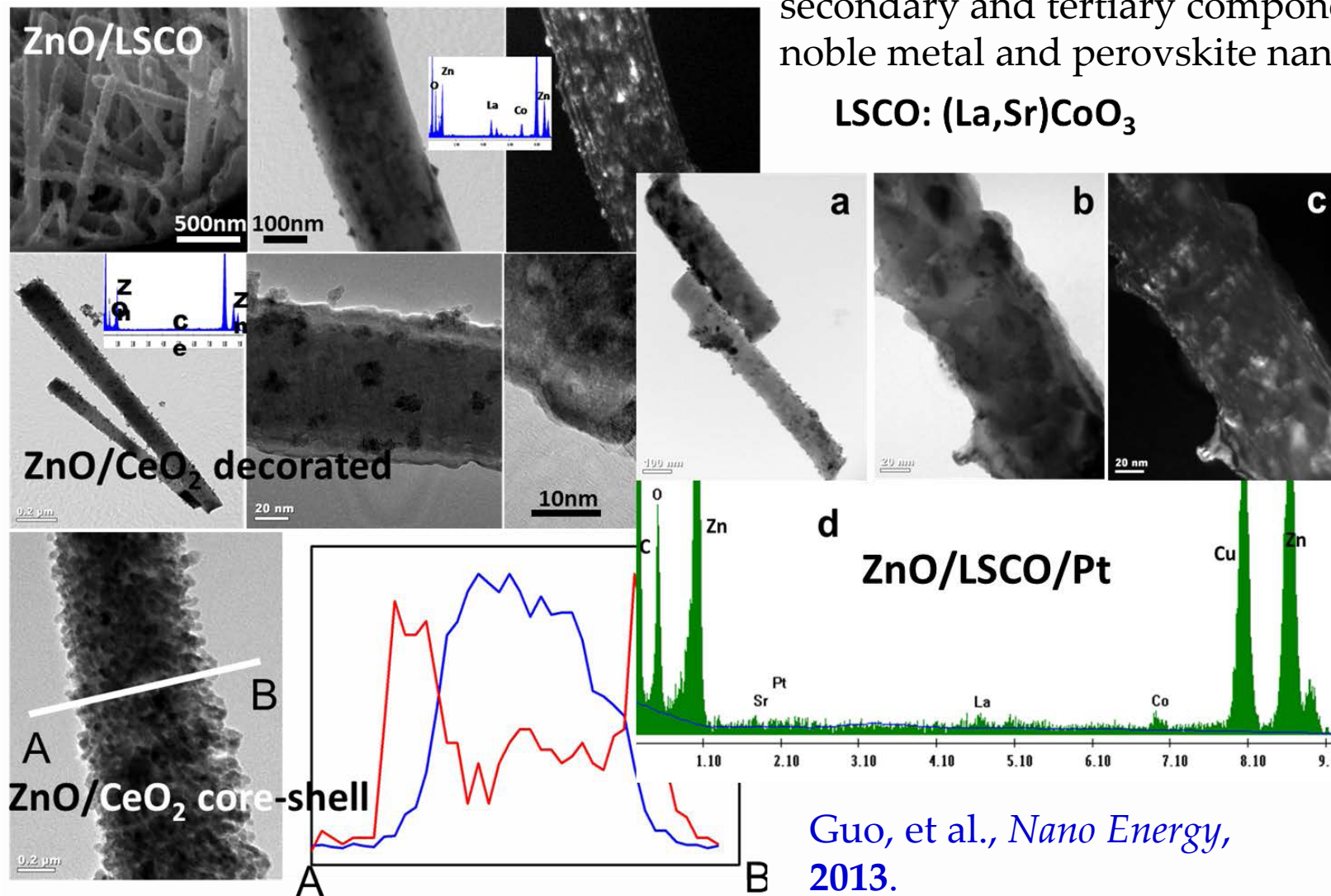
Guo, Ren, Gao et al., *Nano Energy*, **2013**, 2, 873-881.

Ren, Gao et al., *J. Mater. Chem. A*, **2013**, 1, 9897-9906.

Metal Oxide Nano-array based Catalysts

- Solution or vapor phase deposition of the secondary and tertiary components such as noble metal and perovskite nanoparticles.

LSCO: $(\text{La,Sr})\text{CoO}_3$



Guo, et al., *Nano Energy*,
2013.

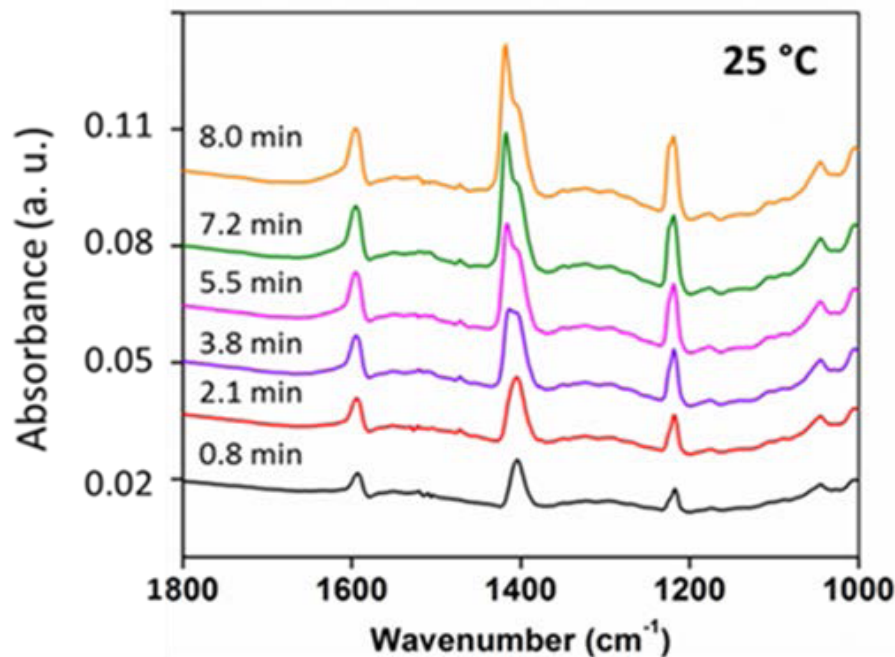
Project Milestones

- **FY15 Quarterly Milestones:**
 - Q1: Synthesize metal oxide nano-array samples for decoration of perovskites
 - complete
 - Q2: Synthesize metal oxide/perovskite nano-array samples for catalytic testing
 - complete
 - Q3: Synthesize metal oxide/perovskite/metal nano-array samples for catalytic testing
 - on track
 - Q4: Characterize structural characteristics of nano-array catalysts in correlation with the catalytic testing performance
 - on track

Collaborations

- Oak Ridge National Laboratory: In-situ spectroscopy characterization of nano-array catalysts with Dr. Zili Wu through Center for Nanophase Materials Science.

- ❖ Doped Co_3O_4 catalyst for low temperature propane oxidation.
- ❖ Controlled Ni doping enhanced reaction kinetics and catalytic activity.
- ❖ A redox reaction mechanism revealed by *in situ* spectroscopy.
- ❖ Declined thermal stability with Ni concentration due to NiO segregation.



Ren, Wu, Gao, et al., *Appl. Catalysis B*, 2015.

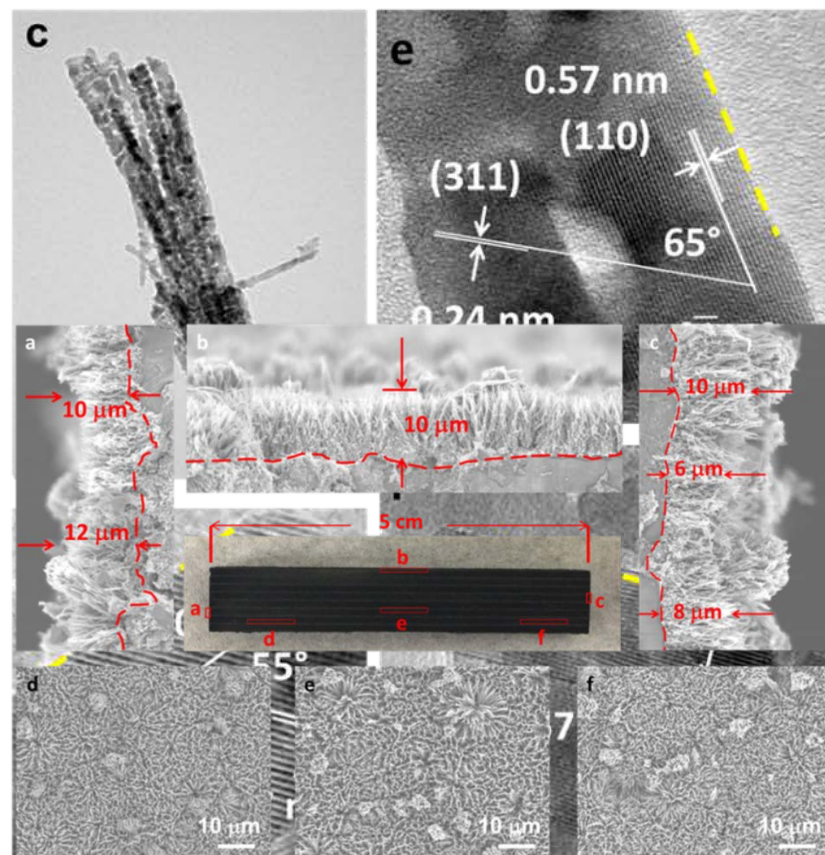
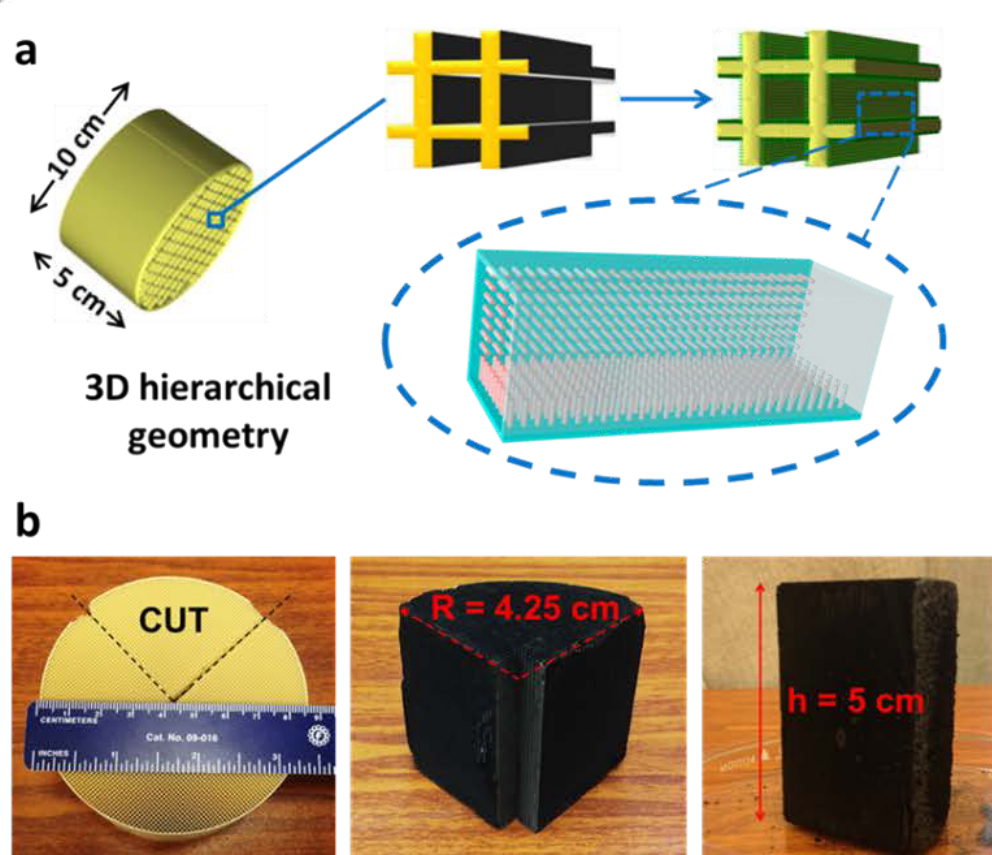
- Brookhaven National Laboratory: metal oxide and metal loading study, and fine structure and chemical analysis of nano-array catalysts with Chang-Yong Nam through Center for Functional Nanomaterials.

Accomplishments

(Project period: 10/1/2014-03/31/2015)

- 1) Synthesis, characterization and testing of PGM free Co_3O_4 based nano-array based monolithic catalysts.
- 2) Synthesis, characterization and testing of PGM free MnO_2 based nano-array based monolithic catalysts.
- 3) Synthesis, characterization and testing of perovskite and Pt nanoparticles loaded metal oxide nano-array catalysts.
- 4) Formulation and initial testing of promising nano-array based monolithic catalysts with low temperature catalytic oxidation performance toward CO and HCs oxidation.

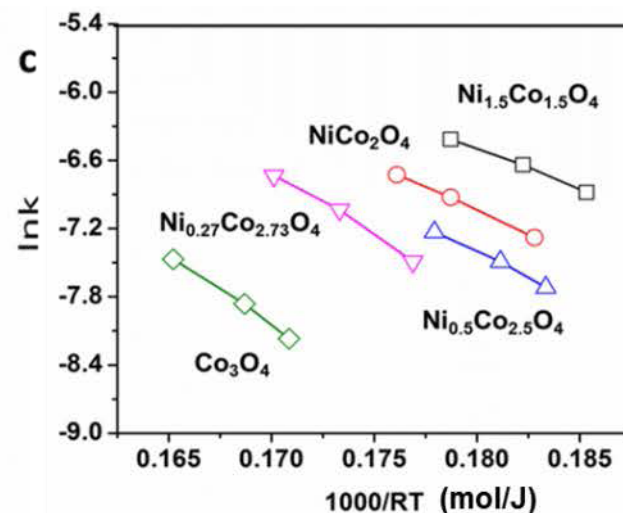
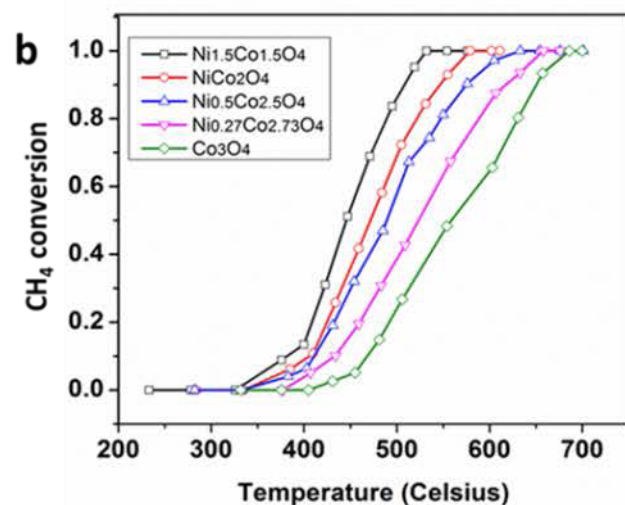
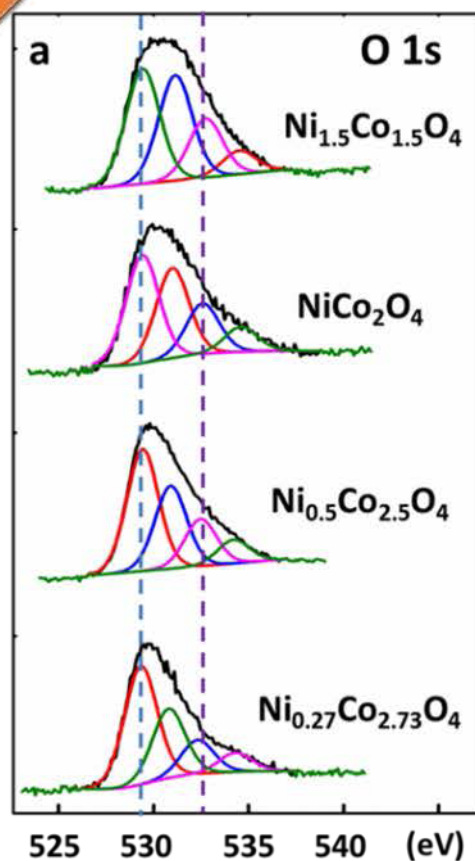
PGM-free Nano-array Catalysts: Spinel $M_x\text{Co}_{3-x}\text{O}_4$ (M=Co, Ni and Zn)



a) Monolithic integration of nano-arrays on commercialized honeycomb supports; b) Photographs of a piece of monolithic nano-arrays catalyst; c) TEM characterization of the Co_3O_4 nanorarrays; HRTEM investigation of d) Co_3O_4 , e) $\text{Ni}_{0.5}\text{Co}_{2.5}\text{O}_4$ and f) $\text{Zn}_{0.5}\text{Co}_{2.5}\text{O}_4$ nano-arrays.

PGM-free Co_3O_4 based Nano-array Catalysts:

Low temperature HC oxidation tunability

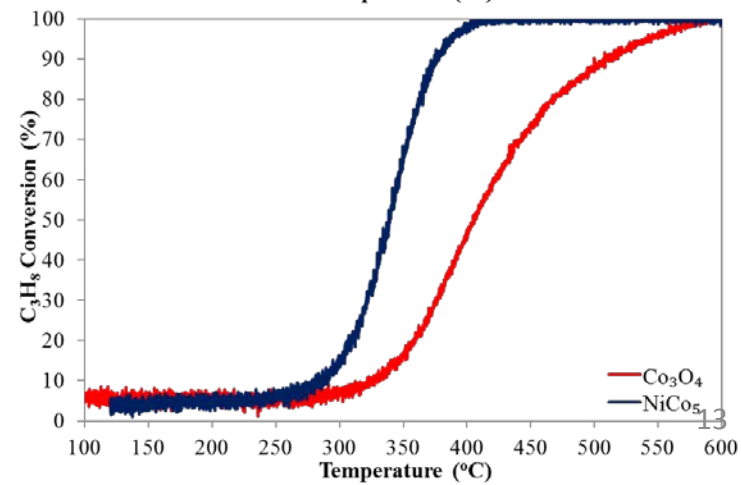
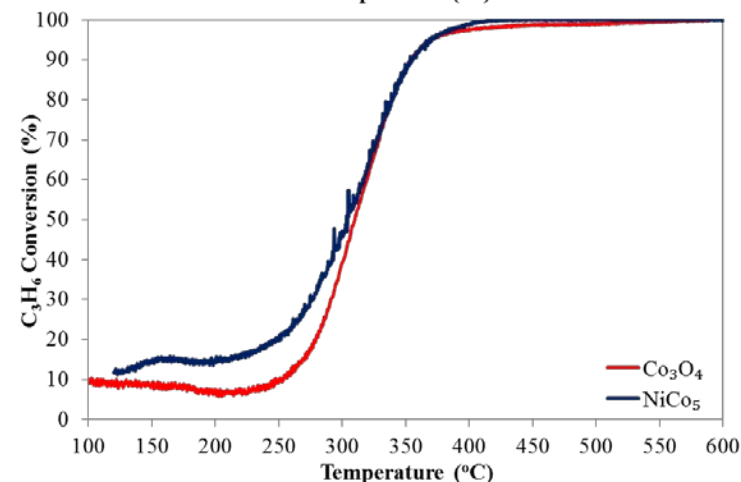
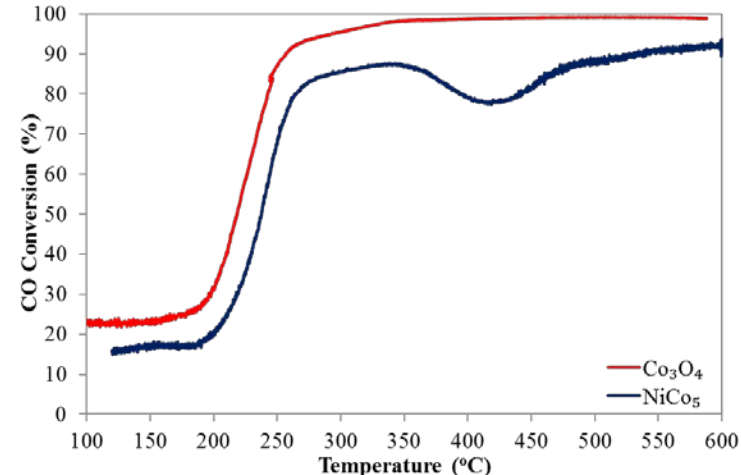
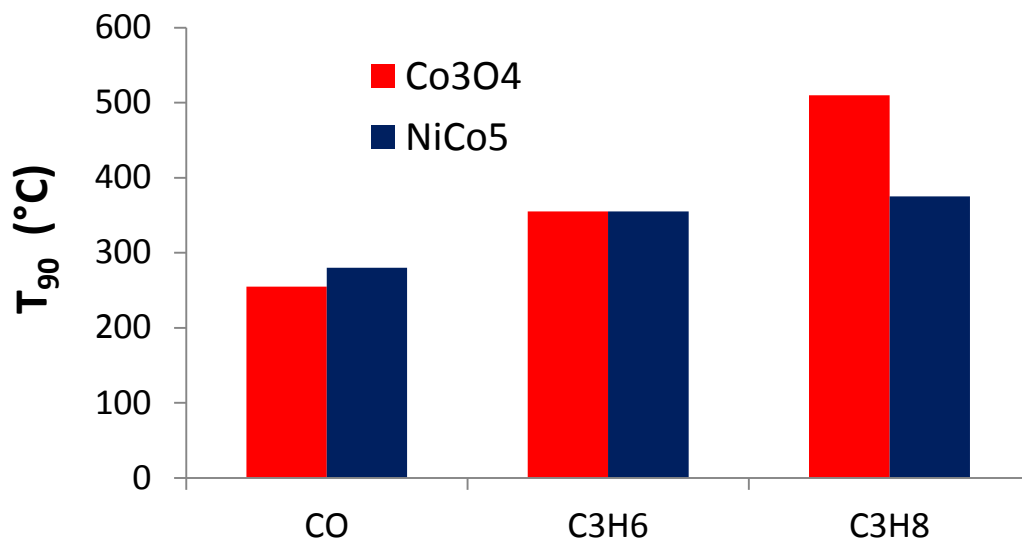


	$\text{Ni}_{1.5}\text{Co}_{1.5}\text{O}_4$	NiCo_2O_4	$\text{Ni}_{0.5}\text{Co}_{2.5}\text{O}_4$	$\text{Ni}_{0.27}\text{Co}_{2.73}\text{O}_4$	Co_3O_4
S_{BET} (m^2/g)	137.8	154.2	146.2	142.7	155.6
E_a (kJ/mol)	70.4	82.6	89.1	111.7	124
T_{50} ($^{\circ}\text{C}$)	440	460	475	525	560
T_{100} ($^{\circ}\text{C}$)	523	560	630	655	680

a) XPS spectra of $\text{Ni}_x\text{Co}_{3-x}\text{O}_4$ with different Ni/Co ratios; b) enhanced methane combustion with higher Ni concentration; c) Arrhenius plots of $\text{Ni}_x\text{Co}_{3-x}\text{O}_4$ for methane combustion; d) Summary of surface area, apparent activation energy and characteristic reaction temperatures of $\text{Ni}_x\text{Co}_{3-x}\text{O}_4$.

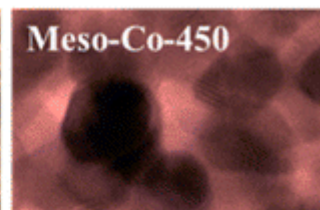
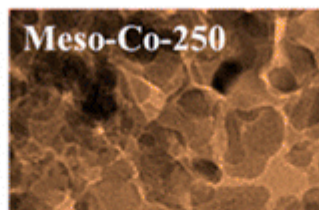
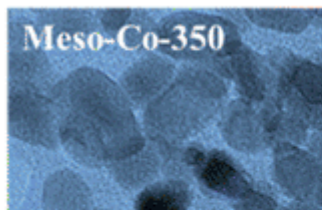
Oxidation Behavior of NiCo_5 and Co_3O_4 Catalysts

- Ni-doped Co_3O_4 (NiCo_5) and Co_3O_4 nano-array catalysts illustrate a range of activities
 - NiCo_5 catalyst better at C_3H_8
 - Co_3O_4 catalyst better at CO and NO to NO_2
- Under conditions analyzed neither meets the goal of 90% conversion at 150°C
 - ...but no PGM and room for improvement

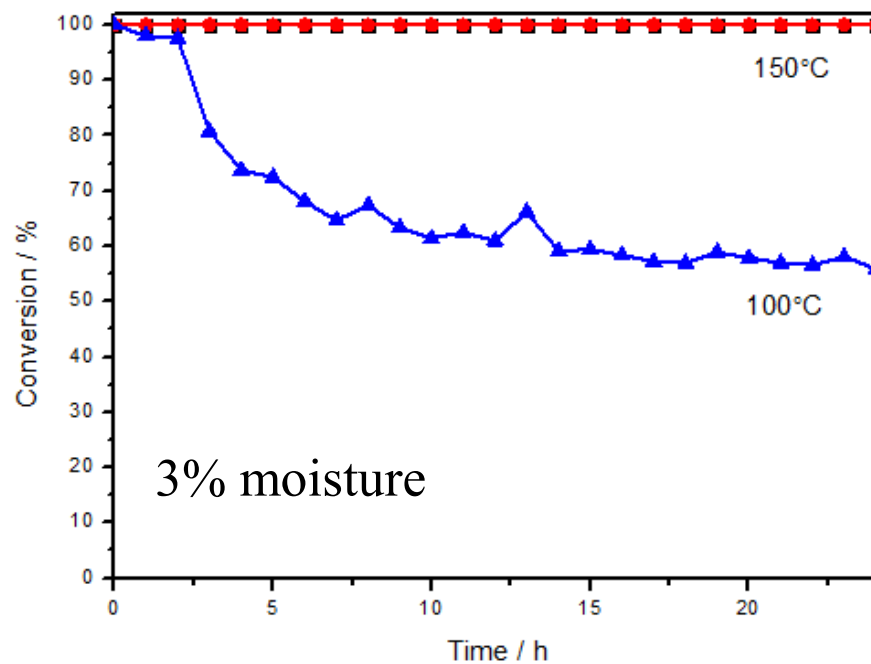
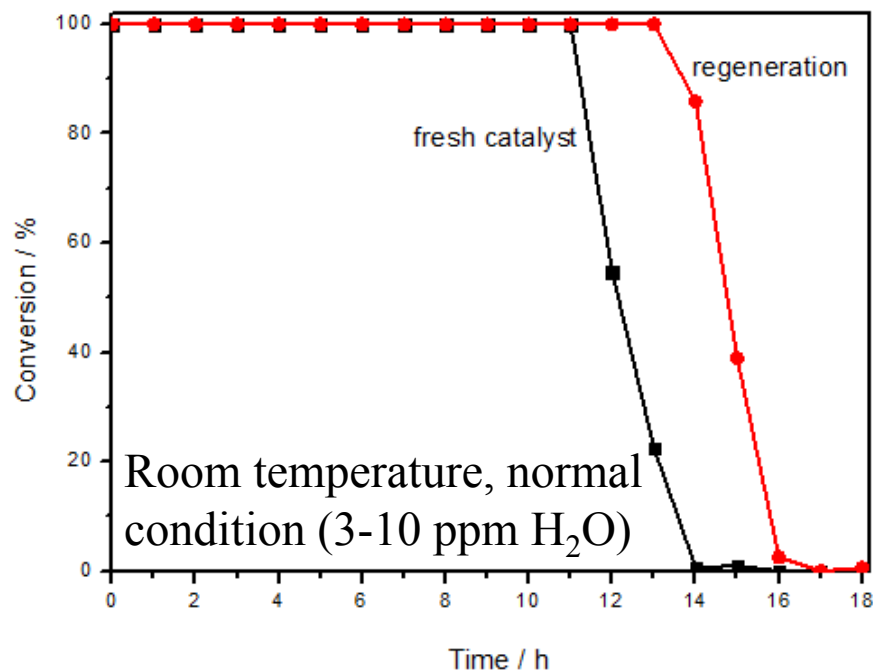


Low Temperature CO Oxidation of Mesoporous Co_3O_4 Nanoparticles

100% CO conversion

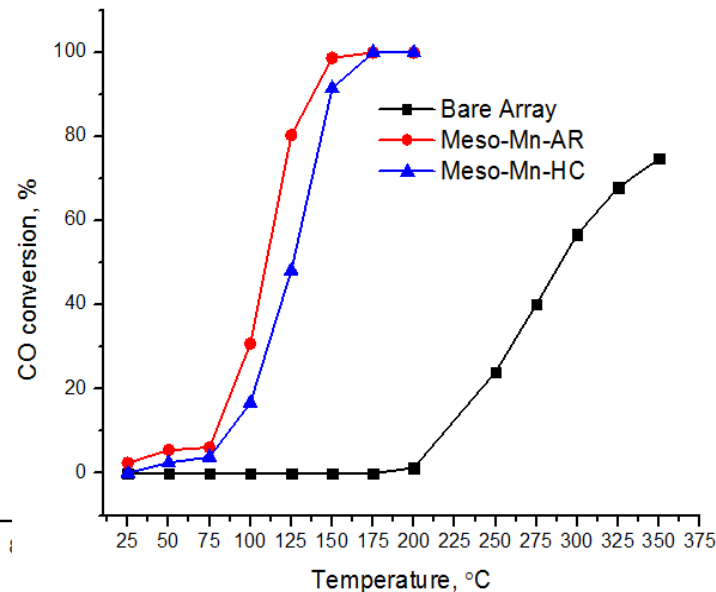
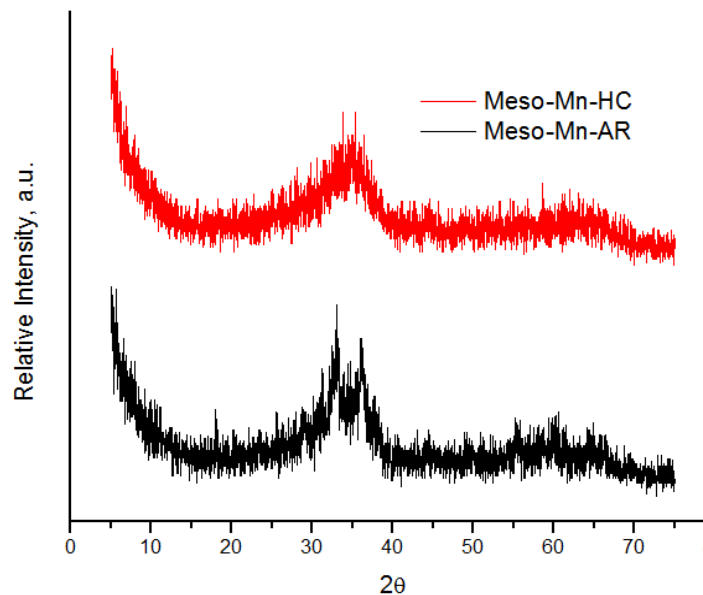
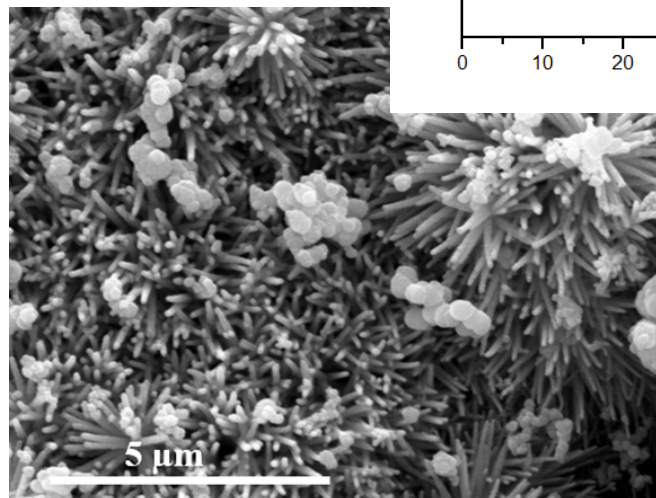
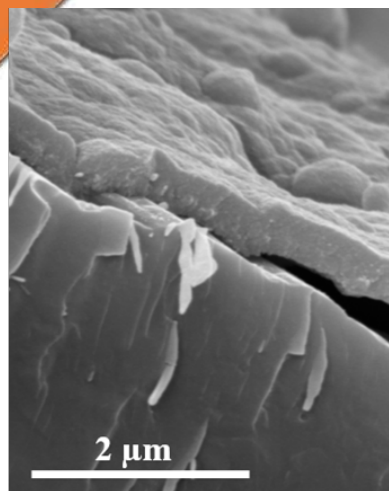


inverse surfactant
micelle synthesis
method



- Low temperature CO oxidation performance of mesoporous Co_3O_4 nanoparticles at normal and moisture condition.

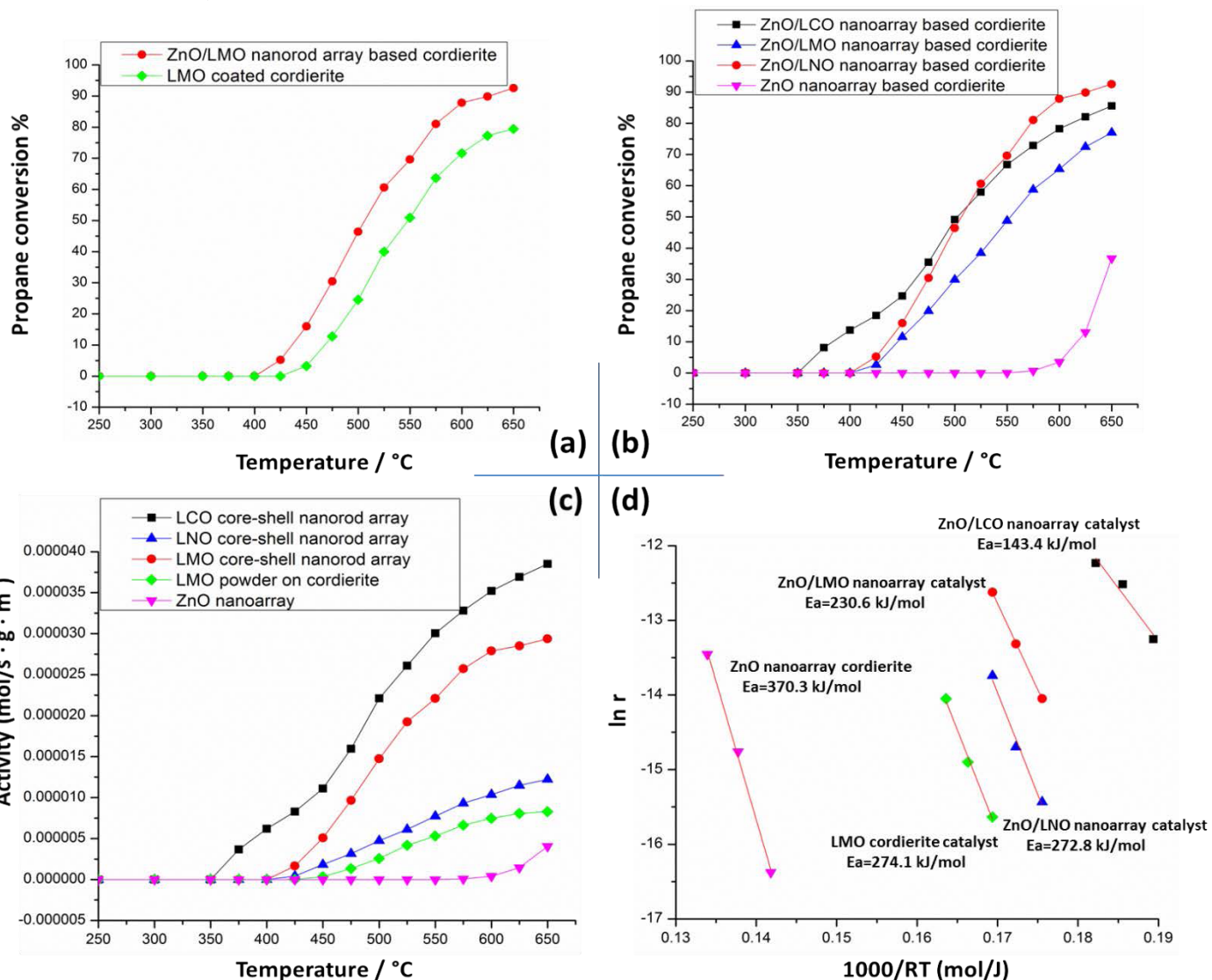
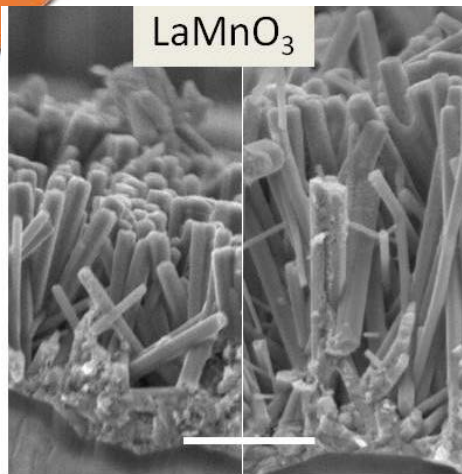
Low Temperature CO Oxidation of Mesoporous MnO₂ loaded Nano-arrays



Material	Loading Ratio	Mass loading	T ₅₀
Meso-Mn-HC	~ 8%	~ 9 mg	126 °C
Meso-Mn-AR	~ 17%	~ 18 mg	110 °C
Bare array	--	--	290 °C

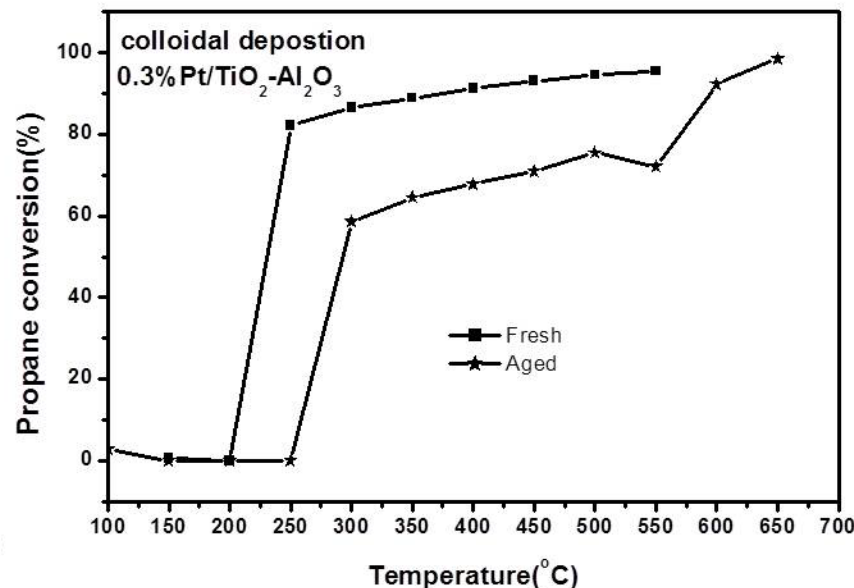
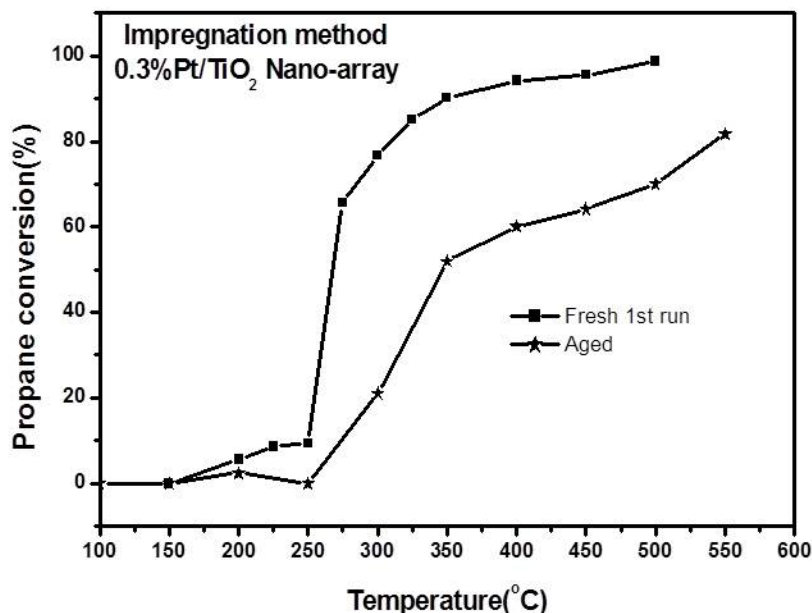
- Low temperature CO oxidation performance of mesoporous MnO₂ decorated MnO₂ nano-arrays at normal condition

Perovskite loaded Nano-arrays: *Propane Oxidation*



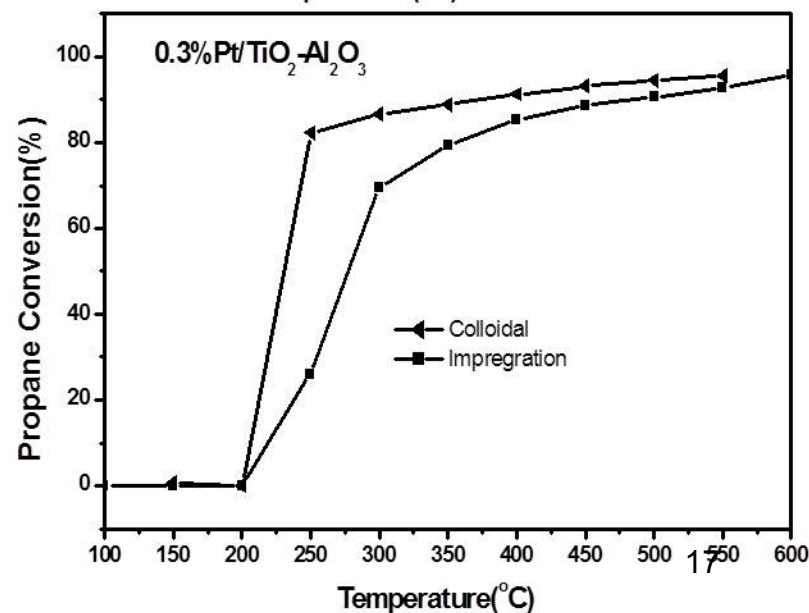
- Interface/loading effects: 25°C lower light-off temperature than wash-coated perovskite catalysts (LMO loading, 4.3mg);
- Composition effect: ZnO/LaBO₃ nano-arrays with catalytic activity sequence of LaCoO₃ > LaMnO₃ > La₂NiO₄ at the initial stage of catalytic reaction

Pt/TiO₂ based Nano-array Catalysts: Propane Oxidation



* Hydrothermal aging: 10% H₂O vapor 800°C, 10 hours

- 0.3 wt.% Pt loaded TiO₂ and TiO₂-Al₂O₃ nano-array monoliths: ~ 80% C₃H₈ conversion at ~250°C. SV: 36,000 h⁻¹
- Colloidal deposition better than impregnation in catalyst performance.
- Hydrothermal aging degrades catalytic performance, with Pt/TiO₂-Al₂O₃ better sustained.



Hoang, Guo, Gao, et al., *unpublished*, 2015.

Future work

- 1) Formulation of selective metal oxide nano-array catalysts with good catalytic oxidation performance at 150 °C or lower.
- 2) Optimized loading of the noble metal and perovskite nanoparticles on selective nano-array catalysts.
- 3) Evaluation of oxidation behavior of nano-array catalysts over CO and HCs oxidation under simulated exhaust atmosphere.
- 4) Assembly of large scale and selective nano-array catalysts for engine testing in FY 16.

Acknowledgements

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- Collaborators: Drs. Z. Wu, S. Overbury, J. Parks (ORNL), Dr. C. Nam, D. Su, H. Chen (BNL)
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